L16 ANSWER 131 OF 260 REGISTRY COPYRIGHT 2007 ACS on STN

RN 186599-61-1 REGISTRY

ED Entered STN: 27 Feb 1997

CN Cobalt, $(2,2'-bipyridine-\kappa N1,\kappa N1')[(1,2,3,4,5-\eta)-1,2,3,4,5-\eta)$ pentamethyl-2,4-cyclopentadien-1-yl]- (9CI) (CA INDEX NAME)

MF C20 H23 Co N2

CI CCS

SR CA

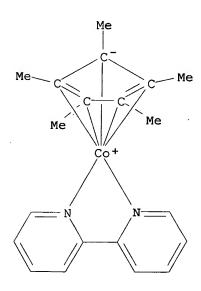
LC STN Files: CA, CAPLUS

DT.CA CAplus document type: Journal

RL.NP Roles from non-patents: FORM (Formation, nonpreparative); PREP (Preparation); PROC (Process); PRP (Properties)

Ring System Data

Elemental Analysis EA	Elemental Sequence ES	Size of the Rings	Ring System Formula RF		RID Occurrence Count
C2Co-C2Co- C2Co-C2Co- C2Co-C2CoN2- C5N-C5N	CoC2-CoC2- CoC2-CoC2-	3-3-3-3-5- 6-6	C15CoN2	4607.1.1	1



- 3 REFERENCES IN FILE CA (1907 TO DATE)
- 3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

AN 142:114242 CA <<LOGINID::20070215>>

TI Reduced and excited states of the intermediates (α -diimine) (C5R5)Rh in hydride transfer catalysis schemes: EPR and resonance Raman spectroscopy, and comparative DFT calculations of Co, Rh and Ir analogues

AU Sieger, Monika; Kaim, Wolfgang; Stufkens, Derk J.; Snoeck, Theo L.; Stoll, Hermann; Zalis, Stanislav

CS Institut fuer Anorganische Chemie, Universitaet Stuttgart, Stuttgart, D-70550, Germany

SO Dalton Transactions (2004), (22), 3815-3821

CODEN: DTARAF; ISSN: 1477-9226 PB Royal Society of Chemistry DT Journal LΑ English 29-13 (Organometallic and Organometalloidal Compounds) CC Section cross-reference(s): 22, 77 The electronic structures of the highly air-sensitive intermediates AB (N-N) (C5Me5)Rh, (N-N = 2,2'-bipyridine (bpy), 2,2'-bipyrimidine (bpym),2,2'-bipyrazine (bpz) and 3,3'-bipyridazine (bpdz)) of hydride transfer catalysis schemes were studied through resonance Raman (rR) spectroscopy and through EPR of the reduced forms [(N-N)(C5Me5)Rh] --. The rR results are compatible with a predominant MLCT character of the lowest excited states [(N-N)(C5Me5)Rh]*, and the EPR spectra of the reduced states reveal the presence of anion radical ligands, (N-N) o-, coordinated by unusually electron rich rhodium(I) centers. The exptl. results, including the assignments of electronic transitions, are supported by DFT calcns. for the model compds. [(N-N)(C5H5)Rh]o/o-N-N = bpy or bpym. The calcns. confirm a significant but not complete mixing of metal and ligand orbitals in the LUMO which still retains about $3/4 \pi * (N-N)$ character. DFT calcns. on (bpy) (C5H5)M and [(bpy)(C5H5)ClM]+, M = Co, Rh, Ir, agree with the exptl. results such as the differences between the homologues, especially the different LUMO of the precursor cations in the case of Co (ightarrowdM) and Rh or Ir $(\rightarrow \pi * (bpy))$. reduced excited state intermediate diimine cyclopentadienyl rhodium hydride transfer; catalysis hydride transfer diimine cyclopentadienyl rhodium complex; EPR resonance Raman spectra DFT diimine rhodium cobalt iridium IΤ Density functional theory (B3LYP; reduced and excited states of intermediates in cyclopentadienylrhodium diimine in hydride transfer catalysis schemes and EPR, resonance Raman spectroscopy, and comparative DFT calcns. of cobalt, rhodium and iridium analogs) IT Charge transfer transition ESR (electron spin resonance) Electronic structure Electronic transition Excited state Hydride transfer catalysts LUMO (molecular orbital) Resonance Raman spectra (reduced and excited states of intermediates in cyclopentadienylrhodium diimine in hydride transfer catalysis schemes and EPR, resonance Raman spectroscopy, and comparative DFT calcns. of cobalt, rhodium and iridium analogs) IT Raman spectroscopy (resonance; reduced and excited states of intermediates in cyclopentadienylrhodium diimine in hydride transfer catalysis schemes and EPR, resonance Raman spectroscopy, and comparative DFT calcns. of cobalt, rhodium and iridium analogs) IT108120-99-6 108121-03-5 138749-28-7 138749-30-1 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); RCT (Reactant); PROC (Process); RACT (Reactant or reagent) (reduced and excited states of intermediates in cyclopentadienylrhodium diimine in hydride transfer catalysis schemes and EPR, resonance Raman spectroscopy, and comparative DFT calcns. of cobalt, rhodium and iridium analogs) 34475-06-4 IT 46930-73-8 64506-26-9 84180-60-9 91742-27-7 93560-38-4 95392-53-3 102211-72-3 108120-95-2 123421-11-4 145638-41-1 178316-74-0 186599-61-1 603136-08-9 603136-09-0 820969-51-5 820969-55-9

820969-68-4

820969-73-1

820969-71-9 820969-72-0

820969-69-5

820969-70-8

RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation, nonpreparative)

(reduced and excited states of intermediates in cyclopentadienylrhodium diimine in hydride transfer catalysis schemes and EPR, resonance Raman spectroscopy, and comparative DFT calcns. of cobalt, rhodium and iridium analogs)

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REFERENCE 2

- AN 132:293866 CA <<LOGINID::20070215>>
- Synthesis, Structure, and Reactivity of [C5Me5CoLL'] Complexes with L = TT Pyridine and L' = Olefin or L-L' = Bipyridine
- AII Lenges, Christian P.; White, Peter S.; Marshall, Will J.; Brookhart, Maurice
- CS Department of Chemistry Venable and Kenan Laboratories, The University of North Carolina at Chapel Hill, Chapel Hill, NC, 27599-3290, USA
- SO Organometallics (2000), 19(7), 1247-1254 CODEN: ORGND7; ISSN: 0276-7333
- PΒ American Chemical Society
- DTJournal

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LA
     English
CC
     29-13 (Organometallic and Organometalloidal Compounds)
      Section cross-reference(s): 75
AB
     The bis(olefin) complex [C5Me5Co(C2H3SiMe3)2] (6) reacts with pyridine at
     22° to give the pyridine olefin complex [C5Me5Co(C2H3SiMe3)(C5H5N)]
            The addition of excess pyridine does not result in the formation of a
     bis(pyridine) adduct. Thermolysis of solns. of 7 in benzene-d6 results in
     catalytic H/D exchange which incorporates deuterium from the solvent into
     coordinated olefin. Pyridine dissociation generates a 16-electron
     intermediate which is formulated as the reactive species. At long times
     deuteration of the coordinated pyridine is also observed via shuttling of
     deuterium from the solvent to the olefin ligand and then to the pyridine.
     The reaction of 6 with pyridine-d5 results in the analog 7-d5.
     Thermolysis of 7-d5 shows preference for H/D exchange at the ortho
     positions of coordinated pyridine, which has been explained by the
     formation of \eta 2-coordinated intermediates. Reaction of
     4-phenylpyridine with 6 yields the analogous pyridine adduct
     [C5Me5Co(C2H3SiMe3)(4-C6H5-C5H4N)] (12), which was characterized by x-ray structural anal. H/D exchange catalysis is also observed using this pyridine
     derivative In a reaction of bipyridine with 6 clean olefin substitution is observed at 22° to generate violet, highly air sensitive
     [C5Me5Co(bipy)] (13). Complex 13 has been characterized by x-ray
     structural anal.
ST
     silylvinyl cobalt cyclopentadienyl complex reaction pyridine bipyridine;
     pentamethylcyclopentadienylcobalt silylvinyl pyridine complex prepn
     hydrogen deuterium exchange; crystal mol structure silylvinyl
     phenylpyridine pentamethylcyclopentadienylcobalt bipyridine complex
TT
     Exchange reaction
         (of hydrogen with deuterium; of silylvinyl
        pentamethylcyclopentadienylcobalt pyridine complexes)
IT
     Crystal structure
     Molecular structure
         (of silylvinyl pentamethylcyclopentadienylcobalt phenylpyridine and
        pentamethylcyclopentadienylcobalt bipyridine complexes)
IT
     186599-61-1P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
         (preparation and crystal structure of)
IT
     264259-98-5P
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation, crystal structure, and hydrogen/deuterium exchange reaction
        of)
IT
     264260-00-6P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation, thermolysis, and hydrogen/deuterium exchange reaction of)
IT
     188827-34-1
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction with pyridine or bipyridine)
     110-86-1, Pyridine, reactions
IT
                                       366-18-7, 2,2'-Bipyridine
     4-Phenylpyridine
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (synthesis, structure, and reactivity of pentamethylcyclopentadienylcob
        alt complexes with pyridine, olefin, or bipyridine)
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REFERENCE 3
ΆN
     126:144381 CA <<LOGINID::20070215>>
ΤI
     Electron transfer and chloride ligand dissociation in complexes
      [(C5Me5)ClM(bpy)]+/[(C5Me5)M(bpy)]n (M = Co, Rh, Ir; n = 2+, +, 0, -): A
     combined electrochemical and spectroscopic investigation
     Kaim, Wolfgang; Reinhardt, Ralf; Waldhoer, Eberhard; Fiedler, Jan
AU
CS
     Institut fuer Anorganische Chemie der Universitaet, Pfaffenwaldring 55,
     Stuttgart, D-70550, Germany
     Journal of Organometallic Chemistry (1996), 524(1-2), 195-202
SO
     CODEN: JORCAI; ISSN: 0022-328X
PB
     Elsevier
DT
     Journal
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LA

English

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CC 29-13 (Organometallic and Organometalloidal Compounds)
Section cross-reference(s): 22, 72, 77

AB In contrast to the rapid and chemical reversible two-e
elimination reaction [(C5Me5)ClM(bpy)]+ + 2e→(C5Me5)M
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AB In contrast to the rapid and chemical reversible two-electron ECE' reductive elimination reaction [(C5Me5)ClM(bpy)]+ + 2e→(C5Me5)M(bpy) + Cl-,
M = Rh or Ir, the analogous cobalt system exhibits two sep. one-electron steps (EC + E' process) with a persistent, EPR-spectroscopically characterized cobalt(II) intermediate [(C5Me5)Co(bpy)]+. Within the series of coordinatively unsatd. homologous species (C5Me5)M(bpy), the cobalt derivative exhibits the smallest and the iridium homolog the largest metal(I)-to-bpy electron transfer in the ground state, as evident from electrochem. potentials and long-wavelength absorption data. A comparison within that homologous series indicates why the rhodium system, with its intermediate position, is most suitable for hydride transfer catalysis.

ST electron transfer chloride ligand dissorn cyclopentadienylcobalt; cobalt rhodium iridium pentamethylcyclopentadienyl bpy chloride; reductive elimination pentamethylcyclopentadienylcobalt rhodium iridium chloride; electrochem potential pentamethylcyclopentadienylcobalt rhodium iridium chloride; ESR pentamethylcyclopentadienylcobalt rhodium iridium chloride bipyridine

IT Cyclic voltammetry
ESR (electron spin resonance)
Electric potential
Electron transfer
IR spectra

(electron transfer and chloride ligand dissociation in complexes pentamethylcyclopentadienylcobalt -rhodium, -iridium chloride bpy complexes by a combined electrochem. and spectroscopic investigation)

IT 90541-04-1 108120-99-6 115565-17-8 138749-27-6 145638-41-1 186599-61-1

RL: FMU (Formation, unclassified); PEP (Physical, engineering or chemical process); PRP (Properties); FORM (Formation, nonpreparative); PROC (Process)

(electron transfer and chloride ligand dissociation in complexes pentamethylcyclopentadienylcobalt -rhodium, -iridium chloride bpy complexes by a combined electrochem. and spectroscopic investigation)

IT 366-18-7, 2,2'-Bipyridine

RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)

(electron transfer and chloride ligand dissociation in complexes pentamethylcyclopentadienylcobalt -rhodium, -iridium chloride bpy complexes by a combined electrochem. and spectroscopic investigation)

RE.CNT 45 THERE ARE 45 CITED REFERENCES AVAILABLE FOR THIS RECORD

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